

## The First and Facile Synthesis of ( $\pm$ ) Syringaresinol

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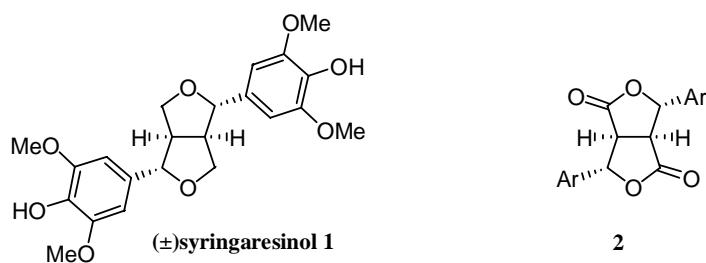
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**Abstract:** The first and facile synthesis of ( $\pm$ )syringaresinol was described.

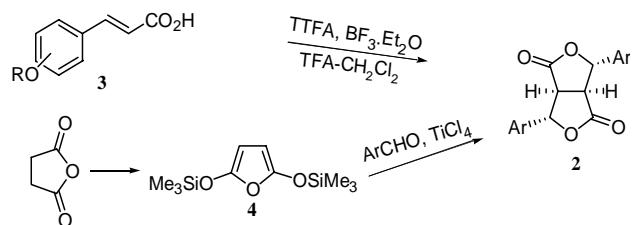
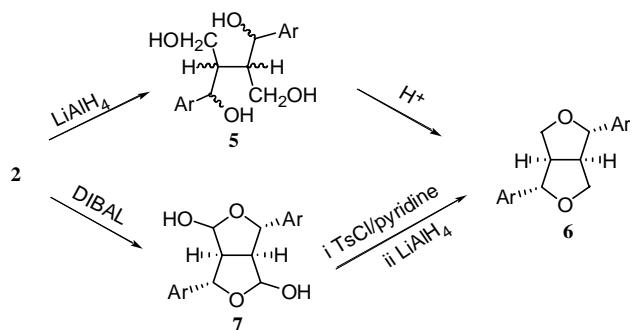
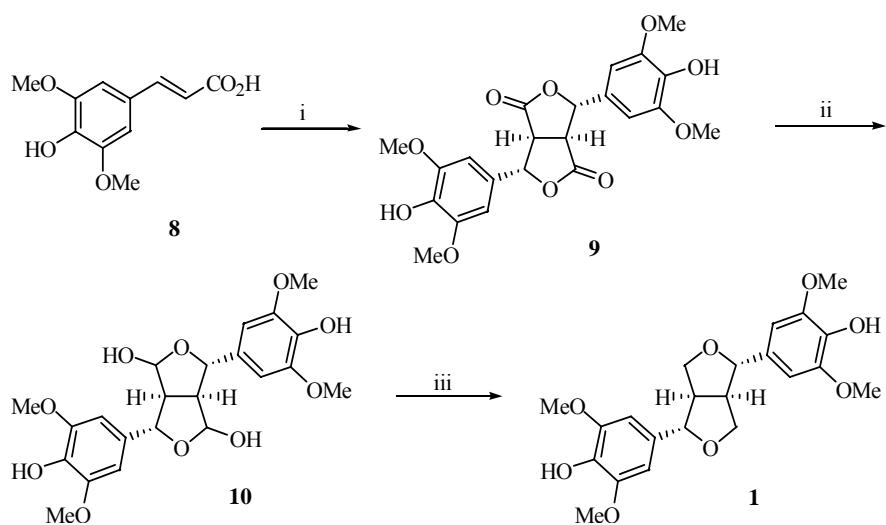
**Keywords:** Lignans, ( $\pm$ )syringaresinol, synthesis.

Lignans have drawn enormous attention of the chemists throughout the world in recent years because of their wide abundance in nature and broad range of biological activities<sup>1</sup>. Syringaresinol **1** and other furofuran lignans have also been reported to exhibit various biological activities, including antifungal<sup>2</sup>, anti-inflammatory<sup>3</sup>, antimalarial activities<sup>4</sup>, inhibition of cyclic AMP phosphodiesterase<sup>5</sup>, inhibiton of platelet aggregation<sup>6</sup>, antileukemic<sup>7</sup>, antioxidation<sup>8</sup> and cytotoxic activities<sup>9</sup>, DNA cleavage effect<sup>10</sup>, etc. Although a number of synthesis of lignans have been reported, there is no report on the synthesis of syringaresinol yet. Herein we like to report the first and facile synthesis of ( $\pm$ )syringaresinol.

For the synthesis of furofuran lignans, usually the 4,8-bislactone **2** was the common intermediate. The general methods so far available for the synthesis of **2** involved the TiCl<sub>4</sub>-catalyzed condensation of 2,5-bis(trimethylsiloxy)furan with aromatic aldehydes<sup>11</sup> or oxidative coupling of cinnamic acid derivatives with thallium trifluoroacetate<sup>12</sup>(TTFA) (**Scheme 1**). However, the former method was not practical, and the latter one needed highly toxic reagent TTFA. The conversion of **2** to furofuran lignans **6** was achieved in low to moderate yield by two approaches, which are illustrated in **Scheme 2**.



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**Scheme 1****Scheme 2****Scheme 3**

Reagents and conditions: i.  $\text{FeCl}_3$ ,  $\text{O}_2$ ,  $\text{EtOH-H}_2\text{O}$ , rt., 54% yield; ii. 1.5eq. DIBAL, THF,  $-78^\circ\text{C}$ , 90% yield; iii. 1.5eq.  $\text{Et}_3\text{SiH}$ , 1.1eq.  $\text{BF}_3\cdot\text{Et}_2\text{O}$ ,  $\text{CH}_2\text{Cl}_2$ ,  $0^\circ\text{C}$ , 88% yield.

Our strategy is to find one straightforward method to produce the target molecule ( $\pm$ )syringaresinol. Thus, oxidative coupling of 4-hydroxy-3, 5-dimethoxycinnamic acid with  $\text{FeCl}_3$  and  $\text{O}_2$  directly gave the bislactone **9**<sup>13</sup>, stereochemistry of which was assigned to be all-*cis* from the small  $H_A$ - $H_B$  coupling constant in the proton *NMR*. Reduction of **9** with DIBAL readily afforded the lactol **10**, and subsequently deoxygenation of **10** with  $\text{Et}_3\text{SiH}$ <sup>14</sup> in the presence of  $\text{BF}_3\text{-Et}_2\text{O}$  produced the target molecule ( $\pm$ ) syringaresinol<sup>15</sup> in high yield (**Scheme 3**). The spectral data of the synthetic sample are in agreement with those reported for (+)- and ( $\pm$ ) syringaresinol<sup>16, 17</sup>

In conclusion, we have described here the first total synthesis of ( $\pm$ ) syringaresinol, and developed a facile method to produce the ( $\pm$ ) syringaresinol straightforwardly.

## References and Notes

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15. The data of synthetic sample **1**: mp. 173-5°C. HREI-MS  $m/z$ : 418.1620[M<sup>+</sup>(C<sub>21</sub>H<sub>26</sub>O<sub>8</sub>), 100%].  
<sup>1</sup>HNMR(CDCl<sub>3</sub>, 400MHz δppm) 6.60 (s, 4H), 5.48(br s, 2H), 4.70(d, 2H,  $J$ = 4Hz), 4.2-4.4(m, 2H), 3.88(s, 12H), 3.8-4.0(m, 2H), 3.08(m, 2H). <sup>13</sup>CNMR(CDCl<sub>3</sub>, 90MHz δppm) 147.26, 134.46, 132.16, 102.86, 86.11, 71.84, 56.42, 54.40.
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